

Table 21.1

$\text{Zn}^{2+}(\text{aq}) + 2\text{e}^{-}$	\rightleftharpoons	$\text{Zn}(\text{s})$	$E^{\ominus} = -0.76 \text{ V}$
$\text{Cr}^{3+}(\text{aq}) + \text{e}^{-}$	\rightleftharpoons	$\text{Cr}^{2+}(\text{aq})$	$E^{\ominus} = -0.42 \text{ V}$
$\text{Ni}^{2+}(\text{aq}) + 2\text{e}^{-}$	\rightleftharpoons	$\text{Ni}(\text{s})$	$E^{\ominus} = -0.25 \text{ V}$
$\text{I}_2(\text{aq}) + 2\text{e}^{-}$	\rightleftharpoons	$2\text{I}^{-}(\text{aq})$	$E^{\ominus} = +0.54 \text{ V}$
$\text{Fe}^{3+}(\text{aq}) + \text{e}^{-}$	\rightleftharpoons	$\text{Fe}^{2+}(\text{aq})$	$E^{\ominus} = +0.77 \text{ V}$
$\text{Cr}_2\text{O}_7^{2-}(\text{aq}) + 14\text{H}^{+}(\text{aq}) + 6\text{e}^{-}$	\rightleftharpoons	$2\text{Cr}^{3+}(\text{aq}) + 7\text{H}_2\text{O}(\text{l})$	$E^{\ominus} = +1.33 \text{ V}$
$\text{H}_2\text{O}_2(\text{aq}) + 2\text{H}^{+}(\text{aq}) + 2\text{e}^{-}$	\rightleftharpoons	$2\text{H}_2\text{O}(\text{l})$	$E^{\ominus} = +1.78 \text{ V}$

